
Chemistry Balancing Equations Worksheet Answers

balancing chemical equations - ap chemistry - balancing chemical equations - answer key balance the equations below: 1) $1 \text{ n}_2 + 3 \text{ h}_2 \rightarrow 2 \text{ nh}_3$ 2) $2 \text{ kclo}_3 \rightarrow 2 \text{ kcl} + 3 \text{ o}_2$ 3) $2 \text{ nacl} + 1 \text{ f}_2 \rightarrow 2 \text{ naf} + 1 \text{ cl}_2$ 4) $2 \text{ h}_2 + 1 \text{ o}_2 \rightarrow 2 \text{ h}_2\text{o}$ 5) $1 \text{ pb}(\text{oh})_2 + 2 \text{ hcl} \rightarrow 2 \text{ h}_2\text{o} + 1 \text{ pbcl}_2$ 6) $2 \text{ albr}_3 + 3 \text{ k}_2\text{so}_4 \rightarrow 6 \text{ kbr} + 1 \text{ al}_2(\text{so}_4)_3$ 7) $1 \text{ ch}_4 + 2 \text{ o}_2 \rightarrow 1 \text{ co}_2 + 2 \text{ h}_2\text{o}$ 8) $1 \text{ c}_3\text{h}_8 + 5 \text{ o}_2 \rightarrow 3 \text{ co}_2 + 4 \text{ h}_2\text{o}$ 9) $2 \text{ c}_8\text{h}_{18} + 25 \text{ o}_2 \rightarrow 16 \text{ co}_2 + 18 \text{ h}_2\text{o}$ 10) $1 \text{ fecl}_3 + 3 \text{ naoh} \rightarrow 1 \text{ fe}(\text{oh})_3 + 3 \text{ nacl}$... **balancing chemical equations worksheet 1** - balancing chemical equations worksheet 2 26. $\text{mg} + \text{cl}_2 \rightarrow \text{mgcl}_2$ 27. $\text{ag}_2\text{o} \rightarrow \text{ag} + \text{o}_2$ 28. $\text{k} + \text{o}_2 \rightarrow \text{k}_2\text{o}$ 29. $\text{cl}_2 + \text{f}_2 \rightarrow \text{clf}_3$ 30. $\text{sio}_2 + \text{c} \rightarrow \text{si} + \text{co}$ 31. $\text{nahco}_3 \rightarrow \text{na}_2\text{co}_3 + \text{co}_2 + \text{h}_2\text{o}$ 32. $\text{zns} + \text{o}_2 \rightarrow \text{zno} + \text{so}_2$ 33. $\text{so}_2 + \text{h}_2\text{s} \rightarrow \text{s} + \text{h}_2\text{o}$ 34. $\text{ba} + \text{o}_2 \rightarrow \text{bao}$ 35. $\text{fe} + \text{o}_2 \rightarrow \text{fe}_2\text{o}_3$ 36. $\text{so}_2 + \text{o}_2 \rightarrow \text{so}_3$... **balancing equations: practice problems - north allegheny** - balancing equations: answers to practice problems 1. balanced equations. (coefficients equal to one (1) do not need to be shown in your answers). **balancing chemical equations worksheet est - kvadilabad** - balancing chemical equations - homework sheet grade 10 science part 1: balance the following chemical equations *note, you may need to work out these balancing equations on extra paper 1. $\text{n}_2 + \text{h}_2 \rightarrow \text{nh}_3$ 2. $\text{s} + \text{o}_2 \rightarrow \text{so}_3$ 3. $\text{hgo} \rightarrow \text{hg} + \text{o}_2$ 4. $\text{zn} + \text{hcl} \rightarrow \text{zncl}_2 + \text{h}_2$ 5. $\text{sicl}_4 + \text{h}_2\text{o} \rightarrow \text{h}_4\text{asio}_4 + \text{hcl}$ 6. $\text{na} + \text{h}_2\text{o} \rightarrow \text{naoh} + \text{h}_2$ 7. h_3po_4 4) $\text{h}_4\text{p}_2\text{o}_7 + \text{h}_2\text{o} \rightarrow \text{si}_2\text{h}_3 + \text{o}_2$ $\text{sio}_2 + \text{h}_2\text{o} \rightarrow \text{al}(\text{oh})_3$... **balancing equations worksheet, part 2 - nlesd** - this work is licensed under a creative commons attribution - noncommercial 4.0 international license. more chemistry tutorials and practice can be found at chemfiesta. **balancing word equations chapter 9 - my chemistry class** - worksheet balancing word equations chapter 10 (remember the following are diatomic: h_2 , n_2 , o_2 , f_2 , cl_2 , br_2 , i_2) the coefficients should add up to the number at the end that is in parenthesis. **worksheet: writing and balancing chemical reactions** - worksheet: writing and balancing chemical reactions 1. balance the following equations and indicate the type of reaction as formation, decomposition, single **balancing equations worksheet - 3-13** - honors chemistry name: _____ writing and balancing equations worksheet sto.1 balance a chemical equation. sto.2 identify the parts of a chemical equation. **chapter 7 worksheet #1 balancing chemical equations** - chapter 7 worksheet #1 balancing chemical equations balance the equations below: 1) ... write the word equations below as chemical equations and balance: 1) zinc and lead (ii) nitrate react to form zinc nitrate and lead. 2) aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas. 3) sodium phosphate and calcium chloride react to form calcium phosphate and sodium ...

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